

O-Sn (Oxygen-Tin)

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The partial Sn-O phase diagram (0-66.7 at.% O) in [Masalski2] was redrawn from [Hansen], which is based on [1949Spa]. Two intermediate phases, Sn_3O_4 and SnO_2 , were shown. In addition, [1974Moh] reported the existence of SnO , which is formed by a (βSn) + $\text{Sn}_3\text{O}_4 \rightarrow \text{SnO}$ peritectoid reaction at 270 °C.

Based on these experimental phase boundary data and available information on thermodynamic properties of this system, [2003Cah] calculated the Sn-O phase diagram. The result is shown in Fig. 1.

References

- 1949Spa:** H. Spandau and E.J. Kohlmeier, The Zinc-Oxygen System, *Z. Metallkde.*, 1949, **40**, p 374-376, in German
1974Moh: G.H. Moh, *Chem. Erde*, 1974, **33**, p 243-275
2003Cah: S. Cahen, N. David, J.M. Fiorani, A. Maître, and M. Vilasi, Thermodynamic Modeling of the O-Sn System, *Thermochim. Acta*, 2003, **403**, p 275-285

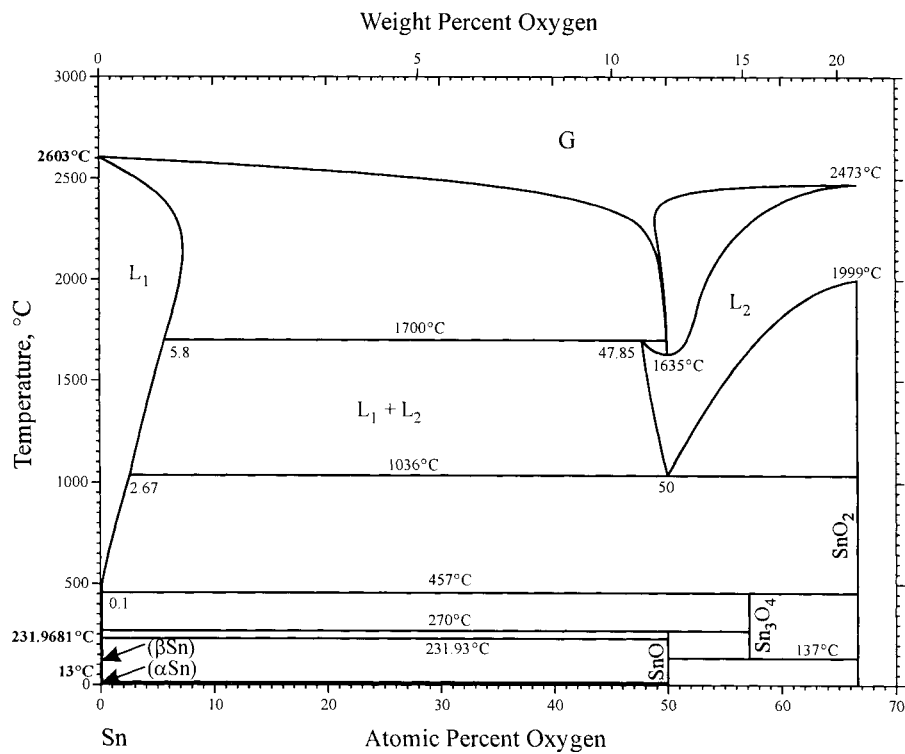


Fig. 1 Sn-O phase diagram